# **AIR MONITORING**

# **DURING THE APPLICATION OF FIRST CONTACT**

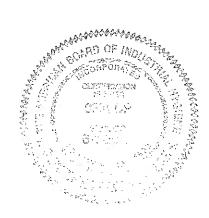
# **ON OPTICAL MIRRORS**

# AT THE LIGO LIVINGSTON OBSERVATORY

IN LIVINGSTON, LOUISIANA

June 2, 2010

TES Job#: SAF1019-10180



Chris T. Robertson

**Certified Industrial Hygienist** 

Certification #: 9304 CP

BY:

TECHNICAL ENVIRONMENTAL SERVICES, INC.

PHONE: 504-348-3098 P.O. BOX 1601 MARRERO, LA 70073 FAX: 504-348-3043



## PURPOSE AND SCOPE SURVEY:

Technical Environmental Services, Inc. (TES) was contracted by LIGO Livingston Observatory (LIGO) to evaluate airborne concentrations of various chemical constituents during the application of a chemical called First Contact. The purpose was to determine employee exposure so that future engineering controls, work practices and personal protection equipment could be administered to protect the employees. All sampling on June 2, 2010 and evaluation of the sample results were completed by Chris Robertson, CIH, CSP.

LIGO LIVINGSTON OBSERVATORY IN LIVINGSTON, LOUISIANA

## SITE BACKGROUND INFORMATION:

LIGO operates a research facility in Livingston, LA. The research calls for the use of 13 inch by 13 inch highly polished optical mirrors. To protect the mirrors, LIGO uses a chemical called First Contact - See the MSDS in **Attachment C**. First contact consists of six distinct hydrocarbons with various concentrations within the formula. The mirrors are housed in a clean room. First Contact is applied using a paint brush, and each application takes approximately four minutes with each coat taking twenty minutes to dry. See **Attachment B** for more information on the application procedure. One employee will be applying the First Contact, and other employees may be in the general area.

During this study the applicator wore a full face respirator with an organic vapor, acid gas, and particulate cartridge, viton gloves, lab coat, hair smock, and shoe covers. The application of the chemical was performed in a 16' x 16' x 16' curtained room. This study involved room sampling both under positive pressure, with 2.5 air changes per minute, and with the positive pressure system turned off. The applicator remained in the room while the chemical was drying.

## SURVEY METHODOLOGY:

3M 3520 passive diffusion monitors were used to collect ethyl alcohol, acetone, and ethyl acetate vapors. Small charcoal tubes in line with SKC sampling pumps were use to collect formal glycol, bis(methoxy)methane, and ethyl lactate. The samples were analyzed at Galson Laboratory, an AIHA certified lab, using the following methods:

> Modified OSHA 7 for formal glycol, bis(methoxy)methane, and ethyl lactate;



- > NIOSH 1400 for ethyl alcohol;
- NIOSH 1300 for acetone; and,
- NIOSH 1457 for ethyl acetate.

See Attachment A for laboratory results.

The breathing zone of the applicator was monitored with both the positive air flow on and the positive air flow off. An area sample was taken outside the containment approximately five feet from the containment, in the breathing zone area, while the positive pressure system was tuned on. The outside area sample was only taken during the worst case scenario, which was with the positive pressure system on. The temperature and humidity were constant at 68°F and 70%, respectively.

#### **SURVEY RESULTS:**

The results of the study revealed very low concentrations of the First Contact chemicals. No overexposures were identified. The results are summarized in **Table I**. The laboratory results can be found in **Attachment A**.

TABLE I: Lab Result and Chemical Exposure Limits

Location/ Person	Agent	Lab Result (ppm)	8-hr ACGIH TLV (ppm)	8-hr OSHA PEL (ppm)	15-min OSHA STEL (ppm)	Target Organ
	Formal Glycol	< 0.13	20	NA	NA	Hematologic
	Bis(methoxy) methane	<0.24	1000	1000	NA	Eye irr, CNS
Applicator	Ethyl Alcohol	4.4	NA	1000	1000	URT irr
w/positive air flow	Acetone	7.3	500	1000	750	URT & eye irr, CNS, hematologic
	Ethyl Lactate	< 0.075	NA	NA	NA	NA
	Ethyl Acetate	<0.7	400	400	NA_	URT & eye irr_

TECHNICAL ENVIRONMENTAL SERVICES, INC. TES Job #:SAF 1019-10180

June 2, 2010 PAGE: 3



# TABLE I - CONT.: Lab Result and Chemical Exposure Limits

Location/ Person	Agent	Lab Result (ppm)	8-hr ACGIH TLV (ppm)	8-hr OSHA PEL (ppm)	15-min OSHA STEL (ppm)	Target Organ
	Formal Glycol	0.64	20	NA	NA	Hematologic
Area	Bis(methoxy) methane	<0.095	1000	1000	NA	Eye irr, CNS
w/positive	Ethyl Alcohol	<0.8	NA	1000	1000	URT irr
air flow	Acetone	1.6	500	1000	750	URT & eye irr, CNS, hematologic
	Ethyl Lactate	< 0.037	NA	NA	NA	NA
	Ethyl Acetate	<0.5	400	400	NA	URT & eye irr
	Formal Glycol	<0.16	20	NA	NA	Hematologic
	Bis(methoxy) methane	<0.23	1000	1000	NA	Eye irr, CNS
Applicator	Ethyl Alcohol	10	NA	1000	1000	URT irr
w/o positive air flow	Acetone	15	500	1000	750	URT & eye irr, CNS, hematologic
	Ethyl Lactate	< 0.065	NA	NA	NA	NA
	Ethyl Acetate	<0.7	400	400	NA	URT & eye irr



#### **ADDITIVE MIXTURE FORMULA:**

The ACGIH suggests that when two or more hazardous substances have a similar toxicological effect on the same target organ, their combined effect, rather that of either individually, should be given primary consideration. This process is known as the Additive Mixture Formula. Based on the very low analytical results, as compared to the respective occupational exposure limits, the additive formula was not applied.

#### **CONCLUSION:**

This limited study revealed very low concentrations of the First Contact chemical agents identified on the MSDS. According to the laboratory results, the ambient concentration of the First Contact chemicals is less within the curtained room with the positive pressure system activated. The outside area concentration was also well below the respective occupational exposure limits. Therefore, respirator protection is not needed if the conditions during the study remain the same and the following recommendations are in place.

## **RECOMMEDATIONS:**

- 1. According to the First Contact MSDS, there are two glove types recommended Viton, and Butyl Rubber. One of these two gloves, should be used during activities that may result in dermal contact with the First Contact chemical. It is not recommended that the client use the Latex class 100 gloves unless permeation studies have be completed that show adequate breakthrough times.
- 2. All containers with liquid chemicals should be covered when not in use. This will help to reduce fugitive emissions from the liquids.
- 3. The applicator should apply the chemical, and then exit the curtain area while waiting the required twenty minutes for the chemical to dry between coats. This will help to reduce any exposure.
- 4. The First Contact chemical should be applied under a laboratory hood or within a positive pressure clean room with a minimum of 2.5 air changes per minute.

## **ATTACHMENTS:**

A: LABORATORY ANALYSIS RESULTS

B: FIRST CONTACT APPLICATION PROCEDURE

C: FIRST CONTACT MSDS



# ATTACHMENT A: LABORATORY ANALYSIS RESULTS



Mr. Chris Robertson Technical Environmental Services, Inc. 5133 Taravella Road Marrero, LA 70072 June 15, 2010

DOH ELAP# 11626

Account# 13258

Login# L215814

Dear Mr. Robertson:

Enclosed are the analytical results for the samples received by our laboratory on June 03, 2010. All test results meet the quality control requirements of AIHA and NELAC unless otherwise stated in this report. All samples on the chain of custody were received in good condition unless otherwise noted.

The samples submitted for 1,3-Dioxolane, Dimethoxymethane and Ethyl Lactate were subcontracted to Bureau Veritas/Clayton Group Services, Inc. Their report is enclosed in its entirety.

Results in this report are based on the sampling data provided by the client and refer only to the samples as they were received at the laboratory. Unless otherwise requested, all samples will be discarded 14 days from the date of this report.

Please contact John Bailey at (877) 482-5227, if you would like any additional information regarding this report.

Thank you for using Galson Laboratories.

Mary & Unangst

Sincerely,

Galson Laboratories

Mary G. Unangst Laboratory Director

Enclosure(s)



6601 Kirkville Road

East Syracuse, NY 13057

(315) 432-5227

www.galsonlabs.com

FAX: (315) 437-0571

Client

: Technical Environmental Services, Inc.

Site : LIGO

Project No.

: SAF 1019-10180

Date Sampled

: 02-JUN-10 Date Received : 03-JUN-10 Account No.: 13258

Login No. : L215814

Date Analyzed : 04-JUN-10 - 05-JUN-10

Report ID

: 649856

Client ID : P01

Lab ID : L215814-13

Time : 60 Minutes

Date Sampled : 06/02/10 Date Analyzed: 06/04/10

<u>Parameter</u>	LOQ uq_	Front uq	Back ug	Total ug	Conc mg/m3	
Acetone	10.	38	<10	42	17	7.3
Ethyl Acetate	5	<5	<5	<5	<3	<0.7
Ethyl Alcohol	5	18	<5	22	8.2	4.4

COMMENTS: Please see attached lab footnote report for any applicable footnotes.

Collection Media

: M3M-3520

Submitted by: JWT

Approved by : KLD

Date: 09-JUN-10 NYS DOH #: 11626

QC by: Tom Burgess

< -Less Than

mg -Milligrams

m3 -Cubic Meters

kg -Kilograms

-Greater Than

ug -Micrograms

1 -Liters

NS -Not Specified

NA -Not Applicable

ND -Not Detected

ppm -Parts per Million LOQ-Limit of Quantitation



6601 Kirkville Road East Syracuse, NY 13057

(315) 432-5227 FAX: (315) 437-0571 www.qalsonlabs.com

Client

: Technical Environmental Services, Inc.

Site : LIGO

Project No.

: SAF 1019-10180

Date Sampled Date Received : 03-JUN-10

: 02-JUN-10

Account No.: 13258 Login No. : L215814

Date Analyzed : 04-JUN-10 - 05-JUN-10

Report ID

: 649856

Client ID : A01

Lab ID : L215814-14

Time: 89 Minutes

Date Sampled: 06/02/10 Date Analyzed: 06/04/10

<u>Parameter</u>	LOQ uq_	Front ug	Back ug	Total ug	Conc mg/m3	
Acetone	10.	12	<10	14	3.9	1.6
Ethyl Acetate	5	<5	<5	<5	<2	<0.5
Ethyl Alcohol	5	<5	<5	<6	<2	<0.8

COMMENTS: Please see attached lab footnote report for any applicable footnotes.

Submitted by: JWT Collection Media : M3M-3520 Approved by : KLD Date : 09-JUN-10 NYS DOH # : 11626 QC by: Tom Burgess

-Less Than mg -Milligrams m3 -Cubic Meters kg -Kilograms NS -Not Specified -Greater Than ug -Micrograms 1 -Liters

NA -Not Applicable ND -Not Detected ppm -Parts per Million LOQ-Limit of Quantitation



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Client

: Technical Environmental Services, Inc.

Site Project No.

: LIGO

: SAF 1019-10180

Date Sampled

: 02-JUN-10

Account No.: 13258

Date Received : 03-JUN-10 Date Analyzed : 04-JUN-10 - 05-JUN-10 Login No. : L215814

Report ID

: 649856

Client ID : P02

Lab ID : L215814-15

Time : 60 Minutes

Date Sampled: 06/02/10

Date Analyzed: 06/05/10

<u>Parameter</u>	LOQ ug_	Front uq	Back uq	Total ug	Conc mg/m3	ppm
Acetone	10.	76	<10	85	35	15
Ethyl Acetate	5	<5	<5	<5	<3	<0.7
Ethyl Alcohol	5	42	<5	50	19	10

COMMENTS: Please see attached lab footnote report for any applicable footnotes.

Collection Media

: M3M-3520

Submitted by: JWT

Approved by : KLD

Date: 09-JUN-10 NYS DOH #: 11626

QC by: Tom Burgess

-Less Than

mg -Milligrams

m3 -Cubic Meters

kg -Kilograms

-Greater Than

ug -Micrograms

1 -Liters NS -Not Specified

NA -Not Applicable

ND -Not Detected

ppm -Parts per Million LOQ-Limit of Quantitation



Client

: Technical Environmental Services, Inc.

6601 Kirkville Road

East Syracuse, NY 13057

(315) 432-5227

FAX: (315) 437-0571 www.galsonlabs.com

Site : LIGO

Project No. : SAF 1019-10180

: 02-JUN-10 Date Sampled

Account No.: 13258 Login No. : L215814

: 03-JUN-10 Date Received Date Analyzed

: 04-JUN-10 - 05-JUN-10

: 649856 Report ID

Client ID : LAB BLANK

Time : NA Lab ID : L215814-16

Date Sampled: Date Analyzed: 06/05/10

<u>Parameter</u>	LOQ uq_	Front ug	Back ug	Total uq	Conc mg/m3	ppm
Acetone	10.	<10	<10	<11	NA	NA
Ethyl Acetate	5	<5	<5	<5	NA	NA
Ethyl Alcohol	5	<5	<5	<6	NA	NA

COMMENTS: Please see attached lab footnote report for any applicable footnotes.

Submitted by: JWT Collection Media : M3M-3520 Approved by : KLD

Date: 09-JUN-10 NYS DOH #: 11626

QC by: Tom Burgess

< -Less Than

mg -Milligrams

m3 -Cubic Meters

kg -Kilograms NS -Not Specified

-Greater Than NA -Not Applicable ug -Micrograms ND -Not Detected -Liters

ppm -Parts per Million LOQ-Limit of Quantitation



6601 Kirkville Road

FAX: (315) 437-0571

www.galsonlabs.com

East Syracuse, NY 13057 (315) 432-5227

#### LABORATORY FOOTNOTE REPORT

Client Name : Technical Environmental Services, Inc.

Site : LIGO

Project No. : SAF 1019-10180

Date Sampled: 02-JUN-10 Account No.: 13258 Date Received: 03-JUN-10 Login No. : L215814

Date Analyzed: 04-JUN-10 - 05-JUN-10

Unless otherwise noted below, all quality control results associated with the samples were within established control limits.

Unrounded results are carried through the calculations that yield the final result and the final result is rounded to the number of significant figures appropriate to the accuracy of the analytical method. Please note that results appearing in the columns preceeding the final result column may have been rounded in order to fit the report format and therefore, if carried through the calculations, may not yield an identical final result to the one reported.

The stated LOQs for each analyte represent the demonstrated LOQ concentrations prior to correction for desorption efficiency (if applicable).

L215814 (Report ID: 649856):

Acetone - Total ug corrected for a desorption efficiency of 90%. Ethyl Acetate - Total ug corrected for a desorption efficiency of 95%. Ethyl Alcohol - Total ug corrected for a desorption efficiency of 84%. SOPs: GC-SOP-12(2), GC-SOP-16(4), GC-SOP-9(3)

Parameter	Method	PEL
Acetone	mod. NIOSH 1300; GC/FID BADGE	1000 ppm
Ethyl Acetate	mod. NIOSH 1457; GC/FID BADGE	400 ppm
Ethyl Alcohol	mod. NIOSH 1400; GC/FID BADGE	1000 ppm

mg -Milligrams ug -Micrograms

-Cubic Meters -Liters

kg -Kilograms NS -Not Specified



June 14, 2010

Shelly Krause GALSON LABORATORIES 6601 Kirkville Road East Syracuse, NY 13057-

Bureau Veritas Work Order No. 10060296

Reference: L215814

Dear Shelly Krause:

Bureau Veritas North America, Inc. received 4 samples on 6/4/2010 for the analyses presented in the following report.

Enclosed is a copy of the Chain-of-Custody record, acknowledging receipt of these samples. Please note that any unused portion of the samples will be discarded 30 days after the date of this report, unless you have requested otherwise.

This material is confidential and is intended solely for the person to whom it is addressed. If this is received in error, please contact the number provided below.

We appreciate the opportunity to assist you. If you have any questions concerning this report, please contact a Client Services Representative at (800) 806-5887.

Sincerely,

Wendy Lesnial

Client Services Representative

cc:

## CASE NARRATIVE

Date: 14-Jun-10

Client:

**GALSON LABORATORIES** 

Project:

L215814

Work Order No 10060296

The results of this report relate only to the samples listed in the body of this report.

Unless otherwise noted below, the following statements apply: 1) all samples were received in acceptable condition, 2) all quality control results associated with this sample set were within acceptable limits and/or do not adversely affect the reported results, and 3) the industrial hygiene results have not been blank corrected.

Please note that there are not enough data points to provide statistical limits.

ANALYTICAL RESULTS

Client: GALSON LABORATORIES

Project: L215814 Work Order No: 10060296

Sample Identification: P 62550

Lab Number: 001A Date Sampled: 6/2/2010

Sample Type Charcoal Tube Date Received: 6/4/2010

Analyst CCR Air Volume (L): 7.5

	Ė	Analytical Res		Reporting Limit	Test	Date
Analyte	(µg)	(mg/m³)	(ppm)	(µg)	Method	Analyzed
1,3-Dioxolane	<3	<0.40	< 0.13	3	OSHA 7	06/06/2010

Sample Identification: A 62549

Lab Number: 002A Date Sampled: 6/2/2010

Sample Type Charcoal Tube Date Received: 6/4/2010

Analyst CCR Air Volume (L): 18

E N	A	Analytical Res	ults	Reporting Limit	Test	Date
Analyte	(µg)	(mg/m³)	(ppm)	(µg)	Method	Analyzed
1,3-Dioxolane	35	1.9	0.64	3	OSHA 7	06/06/2010

Sample Identification: N 62555

Lab Number: 003A Date Sampled: 6/2/2010

Sample Type Charcoal Tube Date Received: 6/4/2010

Analyst CCR Air Volume (L): 6.2

	,	Analytical Res	ults	Reporting Limit	Test	Date
Analyte	(μg)	(mg/m³)	(ppm)	(µg)	Method	Analyzed
1,3-Dioxolane	<3	<0.48	<0.16	3	OSHA 7	06/06/2010

Date: 14-Jun-10

# ANALYTICAL RESULTS

Date: 14-Jun-10

Client:

GALSON LABORATORIES

Project:

L215814

Work Order No: 10060296

Sample Identification: B 62556 BLANK

Lab Number:

004A

Date Sampled: 6/2/2010

Sample Type

**Charcoal Tube** 

Date Received: 6/4/2010

Analyst

CCR

Air Volume (L): NA

	1	Analytical Res	ults	Reporting Limit	Test	Date
Analyte	(µg)	(mg/m³)	(ppm)	(μg)	Method	Analyzed
1,3-Dioxolane	<3		_	3	OSHA 7	06/06/2010

#### General Notes:

Back sections (if applicable) were checked and showed no significant breakthrough unless otherwise noted.

<sup>&</sup>lt;: Less than the indicated reporting limit (RL).

<sup>--:</sup> Information not available or not applicable.

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Need results by 08/18/10. Rush charges are not authorized		Cinnatura			Print Name		Chain of Custody
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IF YOU DO NOT WANT A LABORATORY BLANK ADDED PLEASE CHECK BOX. If blanks are not submitted or box is not checked, our policy states that a laboratory blank will be added for each analyte and it will be charged at normal rate.	mitted or box is not ch	s are not sub	E CHECK BOX. If blank	( ADDED PLEAS be charged at no	IF YOU DO NOT WANT A LABORATORY BLANK ADDED PLEASE CHECK blank will be added for each analyte and it will be charged at normal rate	WANT A LAE	☐ IF YOU DO NOT V blank will be add
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OSHA 7	1,3-Dioxolane		6.2	226-01	6/2/2010		N 62555
OSHA 7	1,3-Dioxolane	-	18.0	226-01	6/2/2010		A 62549
OSHA 7	1,3-Dioxolane		7.5	226-01	6/2/2010		✓ P 62550
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Sampled By: Client	L215814	Project:			Site Name :		886-432-LABS(5227) Fax: 315-437-0571 www.gatsonlabs.com
	Eax No	1770-754-000	7	PRODE NO.		3	Tel: 315-437-5227
	Phone No	2007	2V 555	Dhana Na	2		East Syracuse NY 13857-9
East Syracuse, NY 13057		East Syracuse, NY 13057	East Syracu		New Client? yes [		EM Kirkville Rd
6601 Kirkville Road		6601 Kirkville Road	6601 Kirk		dress		LABORATORIES
Galson Laboratory		Galson Laboratory	Galson L		Check if change		2 IV.
	Invoice To :	Shelly Krause	Shelly	Report To :			



June 11, 2010

Shelly Krause GALSON LABORATORIES 6601 Kirkville Road East Syracuse, NY 13057-

Bureau Veritas Work Order No. 10060294

Reference: L215814

Dear Shelly Krause:

Bureau Veritas North America, Inc. received 4 samples on 6/4/2010 for the analyses presented in the following report.

Enclosed is a copy of the Chain-of-Custody record, acknowledging receipt of these samples. Please note that any unused portion of the samples will be discarded 30 days after the date of this report, unless you have requested otherwise.

This material is confidential and is intended solely for the person to whom it is addressed. If this is received in error, please contact the number provided below.

We appreciate the opportunity to assist you. If you have any questions concerning this report, please contact a Client Services Representative at (800) 806-5887.

Sincerely,

Wendy Lesniak

Client Services Representative

cc:

Novi. MI 48375

## CASE NARRATIVE

Date: 11-Jun-10

Client:

**GALSON LABORATORIES** 

Project:

L215814

Work Order No 10060294

The results of this report relate only to the samples listed in the body of this report.

Unless otherwise noted below, the following statements apply: 1) all samples were received in acceptable condition, 2) all quality control results associated with this sample set were within acceptable limits and/or do not adversely affect the reported results, and 3) the industrial hygiene results have not been blank corrected.

We do not have enough QC data to provide details at this time.

**ANALYTICAL RESULTS** 

Date: 11-Jun-10

Client:

**GALSON LABORATORIES** 

Project:

L215814

Work Order No: 10060294

Sample Identification: P 62553

Lab Number: Sample Type 001A

Charcoal Tube

Analyst

CCR

Date Sampled: 6/2/2010

Date Received: 6/4/2010

Air Volume (L): 6.8

	A	analytical Res	ults	Reporting Limit	Test	Date
Analyte	(µg)	(mg/m³)	(ppm)	(μg)	Method	Analyzed
Dimethoxymethane	<5	<0.74	<0.24	5	OSHA 7	06/11/2010

Sample Identification: A 60366

Lab Number:

002A

Sample Type

Charcoal Tube

Date Received: 6/4/2010
Air Volume (L): 16.9

Date Sampled: 6/2/2010

Analyst

CCR

	A	nalytical Res	sults	Reporting Limit	Test	Date
Analyte	(μg)	(mg/m³)	(ppm)	(μg)	Method	Analyzed
Dimethoxymethane	<5	<0.30	<0.095	5	OSHA 7	06/11/2010

Sample Identification: N 62552

Lab Number:

003A

Sample Type

Charcoal Tube

Analyst

**CCR** 

Date Sampled: 6/2/2010

Date Received: 6/4/2010

Air Volume (L): 6.9

	A	Analytical Res	ults	Reporting Limit	Test	Date
Analyte	(µg)	(mg/m³)	(ppm)	(μg)	Method	Analyzed
Dimethoxymethane	<5	<0.72	<0.23	5	OSHA 7	06/11/2010

# **ANALYTICAL RESULTS**

Date: 11-Jun-10

Client:

**GALSON LABORATORIES** 

Project:

L215814

Work Order No: 10060294

Sample Identification: B 62551

004A

Charcoal Tube

Date Sampled: 6/2/2010

Date Received: 6/4/2010

Analyst

Lab Number:

Sample Type

CCR

Air Volume (L): NA

	A	Analytical Resi	ults	Reporting Limit	Test	Date
Analyte	(µg)	(mg/m³)	(ppm)	(μg)	Method	Analyzed
Dimethoxymethane	<5			5	OSHA 7	06/11/2010

#### General Notes:

--: Information not available or not applicable.

Back sections (if applicable) were checked and showed no significant breakthrough unless otherwise noted.

<sup>&</sup>lt;: Less than the indicated reporting limit (RL).

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Received by LAB:	Relinquished by:	Chain of Custody	Please provide a	Comments:		☐ IF YOU DO NOT WA						✓ B 62551	√N 62552	A 60366	✓ P 62553	Sample Identification	Same day	Next Day by Noon	Next Day by 6pm	2 Business Days	3 Business Days	4 Business Days	ays	Need Results By: (s	www.galsonlabs.com	888-432-LABS(5227)	East Syracuse, NY 13057-9672	6601 Kirkville Rd	LABORATORIES	CALCO	
8 Catterian	A. Costello	Print Name	an uncertainty statement in ac		blank will be added for each analyte and it will be charged at normal rate	NT A LABORATORY BLAN						6/2/2010	6/2/2010	6/2/2010	6/2/2010	Date Sampled	200%	150% Email Results To:	100% Fax Results To:	75%	50%	35% Credit Card No. :	0% Purchase Order No. :	(surcharge) Verbal Authorization	Site Name:		00	New Client? yes		Check if change	Clayton
			cordance with All		be charged at no	ADDED PLEAS						226-01	226-01	226-01	226-01	Collection Medium		; To :	To:			No.:	No.	tion :			Phone No. :				Report To :
11111	100		HA LQAP policy document Sec		ormal rate.	SE CHECK BOX. If blanks are						BLANK	6.9	16.9	6.8	*Air Volume (liters)/ Passive Monitors (Min)		skrause@galsonlabs.com	Email Only Please			C	13258		Pr		888-432-5227	East Syracuse, NY 13057	6601 Kirkville Road	Galson Laboratory	Shelly Krause
	atillis-	Signature	Please provide an uncertainty statement in accordance with AIHA LQAP policy document Section 2A.5.4.3. Need results by 06/11/10. Rush charges are not authorized.			IF YOU DO NOT WANT A LABORATORY BLANK ADDED PLEASE CHECK BOX. If blanks are not submitted or box is not checked, our policy states that a laboratory						Methylal	Methylal	Methylal	Methylal	Analysis Requested		Com	Fax No.			Card Holder Name :			Project: L215814	Fax No.	27 Phone No. :	Y 13057	Road	atory	se Invoice To :
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June 10, 2010

Shelly Krause **GALSON LABORATORIES** 6601 Kirkville Road East Syracuse, NY 13057-

Bureau Veritas Work Order No. 10060298

Reference: L215814

Dear Shelly Krause:

Bureau Veritas North America, Inc. received 4 samples on 6/4/2010 for the analyses presented in the following report.

Enclosed is a copy of the Chain-of-Custody record, acknowledging receipt of these samples. Please note that any unused portion of the samples will be discarded 30 days after the date of this report, unless you have requested otherwise.

This material is confidential and is intended solely for the person to whom it is addressed. If this is received in error, please contact the number provided below.

We appreciate the opportunity to assist you. If you have any questions concerning this report, please contact a Client Services Representative at (800) 806-5887.

Sincerely,

Client Services Representative

cc:

# **CASE NARRATIVE**

Date: 10-Jun-10

Client:

**GALSON LABORATORIES** 

Project:

L215814

Work Order No 10060298

The results of this report relate only to the samples listed in the body of this report.

Unless otherwise noted below, the following statements apply: 1) all samples were received in acceptable condition, 2) all quality control results associated with this sample set were within acceptable limits and/or do not adversely affect the reported results, and 3) the industrial hygiene results have not been blank corrected.

Below is the statistical precision and accuracy information for ethyl lactate by OSHA 7:

Number of samples =34 Recovery % = 87.40 Relative Standard Deviation % =7.48

# ANALYTICAL RESULTS

Client: **GALSON LABORATORIES** 

L215814 Project: Work Order No: 10060298

Sample Identification: P 62547

Lab Number: 001A

Sample Type Charcoal Tube

SAS Analyst Air Volume (L): 8.25

**Analytical Results** Reporting Limit Test Date Analyte (µg) Method Analyzed  $(\mu g)$ (mg/m³) (ppm) Ethyl Lactate <3 < 0.36 3 < 0.075 OSHA 7 06/10/2010

Sample Identification: A 60357

002A Lab Number:

Sample Type **Charcoal Tube** 

Analyst SAS Air Volume (L): 16.9

**Analytical Results** Reporting Limit Test Date Analyte (µg) Method Analyzed (µg) (mg/m³) (ppm) **Ethyl Lactate** <3 < 0.18 3 < 0.037 OSHA 7 06/10/2010

Sample Identification: N 62548

Lab Number:

003A

Sample Type Charcoal Tube

SAS Analyst

Ethyl Lactate

**Analytical Results** Reporting Limit Test Date Analyte Method (µg) Analyzed (µg) (mg/m³) (ppm)

3

< 0.065

< 0.31

<3

Page 19 of 23 Report Reference:1 Generated:15-JUN-10 17:20

06/10/2010

Date: 10-Jun-10

Date Sampled: 6/2/2010

Date Received: 6/4/2010

Date Sampled: 6/2/2010

Date Received: 6/4/2010

Date Sampled: 6/2/2010

Date Received: 6/4/2010

Air Volume (L): 9.6

OSHA 7

# **ANALYTICAL RESULTS**

**GALSON LABORATORIES** 

Project: L215814 Work Order No: 10060298

Sample Identification: B 62554 BLANK

Lab Number:

004A

Date Sampled: 6/2/2010

Date: 10-Jun-10

Sample Type Charcoal Tube

Date Received: 6/4/2010

Analyst

SAS

Air Volume (L): NA

		Analytical Res	ults	Reporting Limit	Test	Date
Analyte	(µg)	(mg/m³)	(ppm)	(µg)	Method	Analyzed
Ethyl Lactate	<3			3	OSHA 7	06/10/2010

#### General Notes:

Client:

Back sections (if applicable) were checked and showed no significant breakthrough unless otherwise noted.

<sup>&</sup>lt;: Less than the indicated reporting limit (RL).

<sup>--:</sup> Information not available or not applicable.

Relinquished by :	Chain of Custody	Please provide an un	Comments:	1	J IF YOU DO NOT WANT A					√ B 62554	N 62548	V A 60357	V P 62547	Identification	Sample							ays	Need Results By: (surcharge)	www.galsonlabs.com	888-432-LABS(5227) Fax: 315-437-0571	Tel: 315-437-5227	5801 Kirkville Rd	LABORATORIES	GALSON	
H. Costello	Print Name	certainty statement in acc		ach analyte and it will b	LABORATORY BLANK					6/2/2010	6/2/2010	6/2/2010	6/2/2010	Date Sampled			m	% Fax Results To:					arge) Verbal Authorization :	Site Name :		Ē	New Cultura yes	or address	Check if change	Files commission or a second s
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ellita	Signature	tion 2A,5.4.3. Need results by 06			IF YOU DO NOT WANT A LABORATORY BLANK ADDED PLEASE CHECK BOX. If blanks are not submitted or box is not checked, our policy states that a laboratory					Ethyl Lactate	Ethyl Lactate	Ethyl Lactate	Ethyl Lactate	Analysis Requested				Fax No.:			Card Holder Name :			Project: L215814	Fax No.		VY 13057	Road	ratory	Invoice To :
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# ATTACHMENT B: FIRST CONTACT APPLICATION PROCEDURE

#### Rev.

## **Document No**

Sheet 1 of 5

# SPECIFICATION

# First Contact Application and Removal Procedure

AUTHOR: Phelps, M.	DATE:04/04/10	Document Change Notice, Release or Approval
		see LIGO DCC record Status

# **Objective**

This document explains the procedure that was developed for applying and removing First Contact(FC) to vertical Advanced LIGO optics in situ. This procedure works equally well for applying First Contact to horizontal optics.

# **Applicable Documents**

T060161 Armandula, H. FTIR and scatter measurements indicate FC leaves no residue on clean optics.

T070280 Technical information from Photonic Cleaning Technologies.

T0900351 Dannenburg, R. FC Peeling and Charging Tests

T1000137 Phelps, M. Drag Wiping and First Contact

#### 3 **Materials**

List of required materials, manufacturer and part number:

- 1. First Contact, Photonic Cleaning Technologies Part# FCL for 1 Liter
- 2. First Contact thinner, Photonic Cleaning Technologies Part#TFCL for 1 Liter
- 3. Clean room gloves, VWR Part #79999-xxx
- 4. Fine PEEK mesh, McMaster Carr Part# 9289T11
- 5. Nylon brushes, Purdy 100% Nylon Brushes from Home Depot
- 6. Flashlight, Stinger from Copquest Part#10-1552-000 OR barlight from Fiber Optic Systems, Part# FSI-060-250
- 7. Kapton tape, McMaster Carr part#7639A75
- 8. Ion gun, Terra Univeral Part#2005-55

# **Application**

- 1. Put on gloves (LIGO approved kind for cleanroom use, see Materials item #3)
- 2. Pour some First Contact(Materials item #1) into a clean glass petri dish.(Fig.1)
- 3. Pour thinner (Materials item #2) or acetone into a small beaker. (Fig. 1)

# First Contact Application and Removal Procedure



Figure 1: To keep from having sticky petri dishes, wrap them in foil, pour in the FC and then discard the foil when finished.

- 4. Situate your flashlight (see Materials item #6 for the <u>flashlight</u>) so that you can see the FC you are applying. Resituate flashlight throughout procedure if you cannot see what you are doing. This is to avoid getting FC too close to the barrel of the optic, and to avoid drips. You can also use a fiberoptic barlight from FiberOptic Systems Inc. This light illuminates the surface much better but is quite large and unwieldy, it needs either an extra person to hold it or a Bunsen burner stand to keep it in place.
- 5. Immerse brush into First Contact. Wipe off excess against side of petri dish. J. Hamilton at Photonics Cleaning recommends Purdy 100% nylon brushes from Home Depot. Only nylon, PEEK or natural hair brushes can be used.

NOTE: If the optic is vertical it is important that the brush does not drip FC onto the barrel, or let it run down to the bottom edge. If there is not too much on the brush, this should not be a problem.

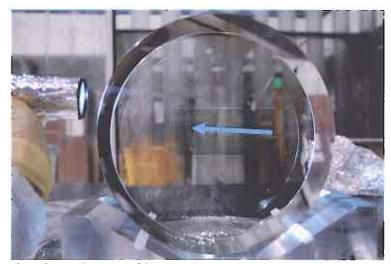


Figure 2: Avoid drips like this one. If it reaches the edge it will be hard to remove. Here the flashlight is used to illuminate but a barlight can be used as well.

# **SPECIFICATION**

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Document No Rev.

Sheet 3 of 5

# First Contact Application and Removal Procedure

- 6. Begin brushing the FC onto the top of the optic. Brush slowly and lightly, avoiding as much bristle to optic contact as possible. Brush FC on in a circle that starts at least 5mm from the optic's edge, taking special care that the FC does not drip or run off the optic face near the bottom. A circle helps keep dripping to a minimum.
- 7. After you complete the initial circle of FC, go back in and fill in the rest using slow, uniform sweeps across the face, as if you were drag wiping.
- 8. Now that one layer is done set your brush in acetone/thinner and wait 20 minutes. First Contact includes acetone, so it is ok to use acetone as a thinner to keep your brush from stiffening. However the thinner sold by Photonic Cleaning works better. Do NOT use methanol as thinner.
- 9. Repeat twice for a total of 3 layers.
- 10. Leave to dry for as long as possible before removing, at least 12 hours.

## 5 Removal

1. Cut a strip of PEEK mesh about 1"x 5". The PEEK mesh is shown here:



Figure 3: Roll of PEEK mesh on aluminum foil

- 2. Hold mesh against the top of the optic. Leave about half the strip sticking up off the top of the optic (Fig.4).
- 3. With one hand hold the PEEK mesh strip, and with your other hand dip your brush into FC and spread a thin layer over the part of the mesh that is on the optic. Be careful to only brush FC onto the layer that is already there. Try not to spread FC all the way to the barrel! It may form a very thin layer, which will be hard to get off. Small strips that get left on near the edge by accident can be wiped off with acetone. Do not use methanol.

# LIGO

# **SPECIFICATION**

Sheet 4 of 5

# First Contact Application and Removal Procedure



Figure 4: Leave mesh sticking up enough to grab it for removal. Note: The mesh in this picture is nylon mesh, not PEEK.

- 4. Let the mesh dry for at least an hour. If the FC film is really stretchy and breaks when you try to remove it, wait longer until it does not stretch when pulled.
- 5. Holding the ion gun in position, begin to peel the PEEK mesh off from the surface. If the dried FC does not come off in one piece STOP, reapply a thicker layer if it has broken apart. As soon as the dry FC film starts coming off, start blowing nitrogen from the ion gun (Materials item #8) onto the optic. Hold the gun parallel to the optic face, focusing most on the area where the film is detaching itself from the optic. Use the flashlight and pay close attention to the edges when peeling it off to make sure nothing is left behind on the optic.



Figure 5: Keep a steady stream of nitrogen while peeling.

6. Once the film is off completely continue blowing off the optic for about 30 seconds to ensure the surface charge is zero. Try to keep the gun at the same angle and avoid waving your arm around,

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Document No Rev.

Sheet 5 of 5

# First Contact Application and Removal Procedure

as you want to avoid disturbing the surrounding particles in the air. You can also point the ion gun towards your gloves before removal, they are plastic and tend to charge up very quickly.

- 7. Shine the flashlight around the outer surface of the optic face to check for any stray dried First Contact. Remove any last bits by wiping a clean lens wipe around the edge in a circular fashion. The middle of the optic will be the cleanest part so wiping from edge to middle to edge will probably drag any contaminants you have into the center.
- 8. Done!

# 6 Alternative Removal(not allowed in vacuum chamber)

- 1. Stick a piece of kapton tape (Materials item #7) against the dried first contact.
- 2. Wait a few seconds, and then pull sharply towards you, the film should come with the tape.
- 3. Do not use this to remove FC in a vacuum chamber where tape is not allowed.

## 7 Additional Information

DO NOT use methanol to drag wipe an optic that has been cleaned with first contact. See <u>First</u> <u>Contact wiki</u> or LIGO #T1000137 for reasons why.

Each layer of FC will take about 20 minutes to "set", ie to be dry enough to allow for the next layer to be applied. For three layers it is good to leave them overnight to ensure it is completely dry. Dry time is longer at cooler temperatures.

If the layer sticks or stretches enough to break while removing it is not dry.

The dried film must leave the optic in one single INTACT piece.

If the dried film is too thin it will break while peeling. If this happens, stop peeling and brush another layer over the dry layer to make a thicker coating and allow re-drying before peeling off again.

When brushing on polymer with a brush do not allow bristles to scour the optic surface.



# **ATTACHMENT C:**

# FIRST CONTACT MSDS

# Safety Data Sheet

According to EC Directive 1907/2006/EC



Date of issue: 19.2.2008 Supersedes edition of: 13.6.2007

## 1. Identification of the substance/preparation and of the company/undertaking

Identification of the product

Product name: First Contact Polymer Solutions

Product numbers: All First Contact Polymer Solution Sizes

Use of the substance/preparation

Chemical for cleaning, protecting, masking and shipping precision surfaces.

Company/undertaking identification

Company:

Photonic Cleaning Technologies, LLC 53818, PO Box 435, Platteville, Wisconsin, USA Tel: +1 608-467-5396, email: safety@photoniccleaning.com

Emergency Telephone No.: Please contact the regional representation in your country (see website).

## 2. Composition/information on ingredients Synonyms: Mix of solvents with inert polymer blend.

CHEMICAL NAMES	CAS NUMBER	MASS CONTENT	EXPOSURE LIMITS IN AIR (UNITS)	
			ACGIH TLV	OSHA PEL
FORMAL GLYCOL	[646-06-0]	10-40%	none est.	none est.
BIS(METHOXY)METHANE	[109-87-5]	10-30%	1000 PPM	1000 PPM
ETHYL ALCOHOL	[64-17-5]	30-50%	400 PPM	400 PPM
ACETONE	[67-64-1]	10-30%	750 PPM	750 PPM
ETHYL LACTATE	[97-64-3]	<10%	none est.	none est.
ETHYL ACETATE	[141-78-6]	<10%	250 PPM	310 PPM

#### 3. Hazards Identification

F, Xi - Highly Flammable, Irritating

#### 4. First aid measures

After inhalation: fresh air.

After skin contact: wash off with plenty of water. Remove contaminated clothing.

After eye contact: rinse out with plenty of water with the eyelid held wide open. Call in ophtalmologist. After swallowing: make victim drink plenty of water, avoid vomiting (risk of aspiration!). No milk. No digestible oils. Laxative: Sodium sulfate (1 tablespoon/1/4 litre water). Immediately call in physician.

## According to EC Directive 1907/2006/EC

Product name: First Contact Polymer Solution

## 5. Fire-fighting measures

Suitable extinguishing media: Water, CO2, foam, powder.

Special risks:

Combustible. Development of hazardous combustion gases or vapours possible in the event of fire. Vapours heavier than air. Could form explosive mixtures with air at ambient temperatures. Beware of backfiring.

Special protective equipment for fire fighting:

Do not stay in dangerous zone without self-contained breathing apparatus.

Other information:

Cool container with spray water from a save distance. Prevent fire-fighting water from entering surface water or groundwater.

#### 6. Accidental release measures

Person-related precautionary measures:

Do not inhale vapours/aerosols. Ensure supply of fresh air in enclosed rooms.

Environmental-protection measures: Do not allow to enter sewerage system; risk of explosion!

Procedures for cleaning / absorption:

Take up with liquid-absorbent material (e.g. Chemizorb®). Forward for disposal. Clean up affected area.

#### 7. Handling and storage

Handling:

Notes for prevention of fire and explosion:

Keep away from sources of ignition. Take measures to prevent electrostatic charging.

Storage:

Tightly closed in a well-ventilated place, away from sources of ignition and heat. Cool (below +25°C).

#### 8. Exposure controls/personal protection

Personal protective equipment:

Protective clothing should be selected specifically for the working place, depending on concentration and quantity of the hazardous substances handled. The resistance of the protective clothing to chemicals should be ascertained with the respective supplier.

Respiratory protection:

required when vapours/aerosols are generated. Filter A (acc. to DIN 3181) for

vapours of organic compounds

Eye protection:

required

# According to EC Directive 1907/2006/EC

Product name: First Contact Polymer Solution

Hand protection:

In full contact:

Glove material:

viton

Layer thickness:

0.70 mm

Breakthrough time:

> 480 in.

In splash contact:

Glove material:

butyl rubber

Layer thickness:

0.7 mm

Breakthrough time:

> 60 min.

The protective gloves to be used must comply with the specifications of EC directive 89/686/EEC and the resultant standard EN374, for example KCL 890 Vitoject® (full contact), 898 Butoject® (splash contact). The breakthrough times stated above were determined for Formyl Glycol (Merck) by KCL in laboratory tests for Merck acc. to EN374 with samples of the recommended glove types. This recommendation applies only to the product stated in the safety data sheet and supplied by us as well as to the purpose specified by us. When dissolving in or mixing with other substances and under conditions deviating from those stated in EN374 please contact the supplier of CE-approved gloves (e.g. KCL GmbH, D-36124 Eichenzell, Internet: www.kcl.de).

## Industrial hygiene:

Immediately change contaminated clothing. Apply skin- protective barrier cream. Wash hands and face after working with substance.

## 9. Physical and chemical properties

Form:

slightly viscous liquid

Colour:

colourless

Odour:

ether-like

pH value

not available circa 100 mPa\*s

Viscosity dynamic (25 °C) Melting Point

above -95 °C

Boiling point

(1013 hPa) below 74 °C

Ignition temperature

unknown °C

Flash point

below 2 °C

Explosion limits

unknown Vol%

upper

unknown Vol%

Vapour pressure (20 °C)

unknown hPa

Relative vapour density

circa 1.7

Density (20 °C)

circa 0.95 g/cm

Solubility in water (20 °C)

partially miscible in small quantities

log Pow:

circa -0.2 (extrapolation from solvents)

lower

## According to EC Directive 1907/2006/EC

Product name: First Contact Polymer Solution

#### 10. Stability and reactivity

Conditions to be avoided

Heating. Upon decomposition in closed containers and tubes risk of bursting due to buildup of overpressure.

Substances to be avoided

Risk of explosion with: perchlorates.

Risk of ignition or formation of inflammable gases or vapors with: fire-promoting substances,

high-oxygen materials / oxidizing agent.

Exothermic reaction with: strong acids, strong bases.

Hazardous decomposition products: no information available

Trace Stabilizer in Formyl Glycol Additive: 2,6-di-t-butyl-4-methylphenol (BHT)

Further information

Explosible with air in a vaporous/gaseous state. light-sensitive, sensitive to air heat-sensitive. Protect from direct sunlight, tends to polymerize (Please observe stabilisation)

## 11. Toxicological information

Acute toxicity for components

LC<sub>so</sub> (formyl glycol) (inhalation, rat): 87 mg/l /4 h.

LD<sub>50</sub> (formyl glycol) (dermal, rabbit): 9040 mg/kg.

LD<sub>50</sub> (formyl glycol) (oral, rat): 3000 mg/kg.

Specific symptoms in animal studies:

Skin irritation test (rabbit): Slight irritations.

Subacute to chronic toxicity for components

Mutagenicity (mammal cell test): negative.

Bacterial mutagenicity: Ames test: negative.

Further toxicological information

After inhalation: drowsiness, headache.

After absorption of large quantities: narcosis.

After skin contact: Slight irritations. Degreasing effect on the skin, possibly followed by secondary inflammation. After eye contact: Slight irritations. After swallowing: gastrointestinal complaints, Nausea, vomiting, diarrhoca.

Further data: The product should be handled with the care usual when dealing with chemicals.

#### 12. Ecological information

Biological degradation:

Solvent Blend Miscible with water. Water-dissolved constituents biodegradable.

Polymer will precipitate when mixed with water.

Behavior in environmental compartments:

Distribution: log p(o/w): -0.37 (experimental for formyl glycol).

No bioaccumulation is to be expected (log  $P(o/w) \le 1$ ).

Henry constant: 2.48 Pa\*m³/mol (formyl glycol). Distribution preferentially in air.

Ecotoxic effects:

Biological effects:

The Safety Data Sheets are also available at www.photoniccleaning.com

## According to EC Directive 1907/2006/EC

Product name: First Contact Polymer Solution

Daphnia toxicity (formyl glycol): Daphnia EC<sub>50</sub>: 6950 mg/l /48 h.

Algeal toxicity (formyl glycol): Selenastrum capricornutum IC<sub>0</sub>: 1000 mg/l /14 d.

Further ecologic data:

No ecological problems are to be expected when the product is handled and used with due care and attention.

## 13. Disposal considerations

Product:

Chemicals must be disposed of in compliance with the respective national regulations. Under www.retrologistik.de you will find country-and substance-specific information as well as contact partners.

Packaging:

*PCT* product packaging must be disposed of in compliance with the country-specific regulations or must be passed to a packaging return system. Under www.retrologistik.de you will find special information on the respective national conditions as well as contact partners.

## 14. Transport information

Road & Rail ADR, RID

Sea IMDG-Code

UN 1170 Ethanol Solution, 3, II

UN 1170 Ethanol Solution, 3, II

Ems F-E S-D

Inland waterway ADN, ADNR not tested

Air CAO, PAX

Ethanol Solution, 3, UN 1170, II

The transport regulations are cited according to international regulations and in the form applicable in Germany. Possible national deviations in other countries are not considered.

## 15. Regulatory information

Labelling according to EC Directives

Symbol: F Highly flammable

R-phrases: 11-36-67 Highly flammable.

S-phrases: 9-16-23-33 Keep away from sources of ignition - No smoking.

EC-No.: 211-463-5 EC label

#### 16. Other information

Reason for alteration

13.6.2007

Updated Version

19.2.2008

Updated Version to EC Directive 1907/2006/EC

Regional representation: see www.photoniccleaning.com

The information contained herein is based on the present state of our knowledge. It characterizes the product with regard to the appropriate safety precautions. It does not represent a guarantee of the properties of the product.